A New Homolytic Method for the Stereospecific Synthesis of (2S,9S)-2-Amino-8-oxo-9,1O-epoxydecanoic Acid in Protected Form

Jack E. Baldwin,^a Robert M. Adlington,^a Christopher R. A. Godfrey^b and Vipulkumar K. Patel^a

a The Dyson Perrins Laboratory, South Parks Road, Oxford OX? 3QY, UK b Jealott's Hill Research Station, Bracknell, Berks RG12 6EY, UK

(2S,9S)-2-Amino-8-oxo-9,1O-epoxydecanoic acid **1, (AOE),** has been synthesized in protected form by homolytic homologation from protected **(S)-2-amino-5-iodopentanoic** acid.

antimitogenic activity. From a synthetic standpoint the epoxy

AOE 1 is the active component¹ of the physiologically active ketone represents a problem because of its high susceptibility family of cyclic tetrapeptides which includes chlamydocin² 2a.

to nucleophilic attack. Consequently, only Schmidt's Structure-activity studies have demonstrated the epoxy syntheses of chlamydocin 2a³ and WF-3161 2b⁴ are of value ketone functionality to be crucial for *in vitro* cytostatic and whereby the epoxy ketone functionality w whereby the epoxy ketone functionality was unmasked after assembly of the incipient AOE residue in a cyclic tetrapeptide

Fig. 1

R2 H

Scheme 1 *Reagents and conditions:* i, Ph3SnH or Bu"3SnH (1 equiv.) (slow addition over $4 h$), cat. AIBN, benzene, reflux; ii $(SiMe₃)₃SiH$ (1.1 equiv.), cat. AIBN, benzene, reflux; AIBN = azoisobutyronitrile

Scheme 2 *Reagents and conditions:* i , $5 \text{ mol\% } (-)$ -DIPT, 6 mol\%
Ti(OPrⁱ)₄, Bu^tO₂H (0.5 equiv.), dried mol. sieves, CH₂Cl₂, -16 °C, 4 $Ti(OPrⁱ)₄$, $Bu^tO₂H$ (0.5 equiv.), dried mol. sieves, $CH₂Cl₂$, h; ii, 5 mol% (+)-DIPT, 6 mol% Ti(OPrⁱ)₄, Bu^tO₂H (0.5 equiv.), dried mol. sieves, CH2C12, -16 "C, **4** h; iii, TFAA (1.5 equiv.), DMSO (2 equiv.), NEt₃ (3 equiv.), CH₂Cl₂, -65 °C then warm to 20 "C; DIPT = diisopropyl tartrate; TFAA = trifluoroacetic anhydride; DMSO = dimethyl sulphoxide

framework. Unfortunately, such a method required multistep sequences *(e.g.* 14 steps from diethyl tartrate to give a precursor to **AOE** in the synthesis of WF-3161 **2b4)** and lacked flexibility. Other reported syntheses of protected **AOE5,6** are not only long but also suffer from not being applicable to the syntheses of the biologically significant tetrapeptide forms.[†] We proposed that a superior approach would be *via* addition of a free-radical **3** derived from a suitably functionalised

t For example, attempted unmasking of any N-protection in order to join onto a tripeptide fragment would result in competing reaction with the sensitive epoxyketone functionality.

cyclopeptide, to a chiral epoxyenone **4,** as a late step in the synthesis (Fig. 1). **As** such, flexibility with respect to the pendant chain length and absolute configuration of the epoxy ketone could readily be accommodated.

In order to test this proposal we firstly attempted a synthesis of **(2S,9S) AOE 5a** in protected form. Initially, protected **(2S)-2-amino-5-iodo-pentanoic** acids **7** and **8,** and (4S)-4,5 epoxypent-1-en-3-one 4 were synthesized.#

Two different methods of radical homologation were then attempted. Slow addition of either tributyltin hydride or triphenyltin hydride (1 equiv.) to a refluxing solution **of** the iodoamino ester **7** (1 equiv.) and epoxyenone **4** (4 equiv.) in benzene gave $(2S,9S)$ *tert*-butyl-N-benzyloxycarbonyl-2**amino-9,1O-epoxy-8-oxo-decanoate, 5a,** in *35%* yield. Using tris(trimethylsily1)silyl hydride7 (1.1 equiv.) as the reducing agent gave similar results (Scheme 1). **A** significant problem encountered in these protocols was decomposition of the enone under the reaction conditions.

Further improvement in the coupling reaction was then demonstrated by the use **of** (4s)- and (4R)-5-trimethylsilyl-**4,5-epoxypent-l-en-3-ones,§ lla** and **llb,** (Scheme **2) as** the radicalophile in the coupling reaction.

Addition of tributyltin hydride (1 equiv.) over 4 h to a refluxing solution of degassed benzene containing protected iodoamino acid (either **7** or **8)** with the trimethylsilyl epoxyenones in fourfold excess gave the terminally silylated **AOEs** in about 65% yield (Scheme 3) (Table 1). Use of tris- (trimethylsily1)silyl hydride as the reducing agent gave variable results ranging from *5* to 60%. The silyl group was then removed by TBAF treatment in DMSO to give both **(9R)-** and **(9s)-AOEs, 5** and **6,** in protected form (Table 1).

 \ddagger Details will appear elsewhere.

[§] The desilylated Mosher's esters8 of the corresponding alcohols, **10a** or **lob,** were shown to be essentially homochiral by 19F NMR analysis.

 $Z = PhCH₂OCO$

Scheme 3 *Reagents and conditions:* i, reagent 11a, Buⁿ₃SnH (slow addition over 4 h), cat. AIBN, benzene, reflux; ii, reagent 11b, Bu3"SnH (slow addition over 4 h), cat. AIBN, benzene, reflux; iii, TBAF (1.1 equiv.), DMSO, 10 min, 20 "C; TBAF = tetrabutylammonium fluoride

Fig. 2 CD spectra **of 5a, 6a, 5b, 6b** and chlamydocin **(2a)**

Circular dichroism (CD) spectra of (2S, 9s) **AOEs 5a** and **6a** showed Cotton effects of the same sign and magnitude as natural chlamydocin **2a9** whereas (2S, *9R) AOEs 5b* and **6b** showed opposite Cotton effects (Fig. 2).

In summary we have developed two related homolytic approaches for the facile assembly of both *(2S,* 9s) and **(2S, 9R)-2-amino-8-oxo-9,lO-epoxydecanoic** acids in protected form. The methodologies are, in principle, suitable for the total synthesis of chlamydocin **2a** and related homologues. Such approaches are current objectives.

Financial support from ICI and the SERC for a **QUOTA** award to V . \hat{K} . P. is gratefully acknowledged. We thank Sandoz Pharma **AG,** CH-4002 Base1 for an authentic sample of chlamydocin **2a** and Dr **A.** Rodger for help in measuring CD spectra.

Received, 3rd June 1991; Corn. 1102635F

References

- 1 **R.** Shute, M. Kawai and D. Rich, *Tetrahedron,* 1988, **44,** 685.
- 2 A. Closse and R. Huguenin, *Helv. Chim. Acta,* 1974, 57, 533.
- 3 U. Schmidt, T. Beuttler, A. Lieberknecht and **H.** Griesser, *Tetrahedron Lett.,* 1983,24, 3573.
- 4 U. Schmidt, U. Beutier and A. Lieberknecht, *Angew. Chem., Int. Engl.,* 1989, **28,** 333.
- 5 R. Jacquier, **R.** Lazaro, H. Raniriseheno and P. Viallefont, *Tetrahedron Lett.,* 1984, *25,* 5525.
- 6 S. Ikegami, T. Hayama, T. Katsuki and M. Yamaguchi, *Tetrahedron Lett.,* 1986, 27,3403.
- 7 B. Geise, B. Kopping and C. Chatgilialogiu, *Tetrahedron Lett.,* 1989, 30, 5479.
- *8* J. Dale, D. Dull and H. Mosher, *J. Org. Chem.,* 1969, 34, 1543.
- 9 M. Kawai, J. Gardner and D. Rich, *Tetrahedron Lett.,* 1986, *27,* 1877.